



Evaluation of *in vitro* Antifungal Activity of Silver and Selenium Nanoparticles against *Alternaria solani* Caused Early Blight Disease on Potato

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Authors' contributions

This work was carried out in collaboration between all authors. Authors AIEB, AWAI and NMS wrote the protocol. Authors AIEB, AWAI and RMF designed and analyzed the present study. Authors AIEB, NMS and RAA wrote the first draft of the manuscript and shared in analysis of the study. Author RMF managed the literature searches, performed the practical studies and statistical analysis. All authors read and approved the final manuscript.

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ABSTRACT

Aim: This study investigated the effect of silver and selenium nanoparticles on *Alternaria solani*, the pathogenic fungus causing early blight disease of potato.

Place and Duration of Study: Drug Radiation Research Department, National Centre for Radiation Research & Technology (NCRRT), Atomic Energy Authority, Cairo, Egypt, 2013.

Methodology: The fungus was isolated from infected potato leaves that showed brown circular spots as early blight disease symptoms. Silver nanoparticles (AgNPs) were prepared biologically using gamma irradiated *Trichoderma viride* cell free supernatant. Selenium nanoparticles (SeNPs)

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were prepared by glutathione method. The synthesized AgNPs were characterized by UV-Vis spectroscopy, Dynamic Light Scattering (DLS) and Transmission Electron Microscope (TEM). Kocide® fungicide was used as reference.

Results: The fungus isolated of leaf spot was identified both microscopically and genetically as *Alternaria solani* causing early blight disease of potato. AgNPs were spherical in shape with average size of 12.7 nm. Selenium nanoparticles were prepared by glutathione as reducing agents. Under laboratory conditions, 25 µg/ml concentration of silver nanoparticles completely inhibited *A. solani* as compared to Kocide® fungicide that gave maximum inhibition at 600 µg/ml. The selenium nanoparticles completely inhibited the fungal growth at 800 µg/ml.

Conclusion: AgNPs completely inhibited the growth of *A. solani* at low concentrations. Silver nanoparticles might be suitable alternative to chemical fungicides. While, SeNPs can be used as antioxidant for enhancing plant immunity.

Keywords: Potato; *Alternaria solani*; early blight; silver and selenium nanoparticles; antifungal.

1. INTRODUCTION

Potato is the fourth main food crop in the world. The economical outlets for this crop are great; however, numerous either soil- or airborne diseases can cause huge losses in the production. Early blight caused by *Alternaria solani* is one of the most serious biotic threat to potato production worldwide causing severe yield losses in Egypt [1].

This pathogen has the ability to grow over a wide range of temperatures [2] and is reported to infect tomato and potato plants under both dry and wet conditions [3], and its propagules survive between crops as mycelia or conidia in soil, plant debris and seed [4].

Values in the literature for measured yield losses due to early blight vary within large limits from 5-78% [5,6].

Diagnosis of a disease in its very early stage can play important role in successful control. Nanotechnologists look for a nano solution for protecting the cultivated plants from bacteria, fungus and viral agents. New molecular and cellular biology tools are expected to provide appropriate pest management practices in agriculture [7].

There are three main methods of synthesizing metal nanoparticles: Chemical, physical and biological methods. The radiation-induced AgNPs synthesis is a simple, clean which involves radiolysis of aqueous solution that provides an efficient method to reduce metal ions [8].

The irradiation, as a new method, had been extensively used to prepare nano-scale clusters

and materials. Radiation induced reduction synthesis of AgNPs which offers some advantages over the conventional methods, because of its simplicity. It provides metal nanoparticles in fully reduced, highly pure and highly stable state [9].

Antifungal activity of ionic or nanoparticle silver has a great potential for use in controlling spore-producing fungal plant pathogens. Silver may be less toxic to humans and animals than synthetic fungicides. Multiple modes of action targeting a broad range of biological pathways of microbes provide an important benefit for avoiding the development of resistance, which has been increasingly important in terms of current issues for the chemical management of many plant fungal diseases [10].

Selenium is an essential nutrient for humans and animals to form important selenoproteins, including glutathione peroxidase, thioredoxin reductase [11]. Selenium has not been classified as an essential element for plants, although its role has been considered to be beneficial in plants capable of accumulating large amounts of the element [12].

2. MATERIALS AND METHODS

2.1 Isolation and Identification of the Pathogenic Fungus, *A. solani*

Standard isolation technique was followed to obtain *A. solani* isolate. For this purpose potato plants showing typical early blight symptoms were collected. The infected leaves were cut into small bits measuring about 2 mm and surface sterilized in 0.1 per cent sodium hypochlorite solution for 2 min. Such bits were transferred into

Petri-dishes containing 15 ml Potato Dextrose Agar (PDA) and incubated at $28 \pm 1^\circ\text{C}$ for 7 days [13]. Pure culture of the fungus was obtained by hyphal tip isolation method [14]. *A. solani* was maintained in the refrigerator on PDA medium and subcultured periodically at an interval of 30 days during the course of this study. The pathogen was identified on the basis of colony characters, viz., colour, growth, pigmentation etc. and morphological characteristics of its mycelium and conidia produced on host and in culture under electron microscope. The fungus was further identified genetically. The partial sequences of 18S rRNA obtained from *Alternaria* isolate was aligned with the available 18S rRNA sequences in Gene Bank data base.

2.2 Biosynthesis of Silver Nanoparticles

Trichoderma viride fungus (provided by Drug Microbiology lab, NCRRT) was used to synthesize silver nanoparticles. The fungus was inoculated in sterilized medium, glucose yeast peptone (GYP) broth containing yeast extract 0.3%, glucose 1%, peptone 0.5%, at 25°C , in shaking condition (180 rpm). After 72 h of incubation, the culture was centrifuged at 5000 rpm for 10 min and the supernatant was used in synthesis of silver nanoparticles [8]. Aqueous solution of AgNO_3 (5 ml) (1 mM AgNO_3 of final concentration) was mixed with fungal supernatant (5 ml) then irradiated at different gamma radiation doses, 0.5, 1.0, 1.5, 2.0, 2.5, 3.0, 5.0, 10, 15, 20, 25 and 30 kGy. AgNPs were characterized by UV-Visible Spectroscopy (T60 UV/Vis spectrometer), Dynamic Light Scattering (DLS) and Transmission electron microscope (TEM).

2.3 Preparation of Selenium Nanoparticles

Selenious acid (0.04 mM) under stirring was mixed with (0.2 mM) glutathione (GSH) as product from *Saccharomyces cerevisiae* (provided by Drug Microbiology lab, NCRRT) and 200 mg bovine albumin solution in 100 ml deionized water. The pH of the mixture was adjusted to 7.2 with 1.0 M sodium hydroxide to initiate the reaction. The reaction lasted one hour under sonication condition, during which the red elemental Se and oxidized glutathione (GSSG) formed. The red solution was dialyzed against doubly distilled water for 96 h with the water changing every 24 h to separate GSSG from Nano-Se [15,16].

2.4 In vitro Comparison Fungicidal Activity between Kocide®, Silver and Selenium Nanoparticles

Inhibitory effect of Kocide®, silver and selenium nanoparticles were tested *in vitro*. A laboratory study was performed to examine the sensitivity of the pathogenic fungus *A. solani* in relation to Kocide®. Nine concentrations of Kocide® were used as follows: 50, 100, 150, 200, 250, 400, 500, 600 and 800 ppm. Six concentrations of silver nanoparticles were used as follows: 5, 10, 15, 20, 25 and 50 ppm respectively. Concentrations of selenium nanoparticles were used as follow: 50, 100, 200, 300, 400, 500, 600, 700 and 800 ppm, respectively. The required concentrations were obtained by adding the appropriate amount of stock solution used, to 100 ml portions of autoclaved potato dextrose agar medium (PDA) which was cooled to about 45°C (not solidified), three Petri dishes were used as replicates for each concentration and three were left without any treatment as control. After solidification of the medium, each dish was inoculated centrally with a mycelial disk (5 mm in diameter) taken from the cultures of the fungus (7 days – old). Plates were incubated at 28°C and colony diameters were measured till untreated control had just covered the plate. Linear growth was measured daily in mm. The average of two perpendicular diameters were recorded. The percentage of inhibition (I %) was calculated according to [17] as follows:

$$I \% = (A - B) / A \times 100$$

Where:

I % = Inhibition percentage.

A = Mean diameter of growth in the control.

B = Mean diameter of growth in treatment.

IC_{50} of the each treatment was determined (IC_{50} , the concentration of each treatment that is required to cause fungitoxic inhibition of the tested fungus by 50 % in comparison with the control).

3. RESULTS AND DISCUSSION

3.1 Isolation and Identification of the Pathogen

According to Mayee and Datar [18], the early blight disease of potato was characterized by the appearance of brown to dark brown colour necrotic spots. Appearance of concentric rings

inside the spots produced target board effect. Singh et al. [19] reported that, the spots were oval to angular in shape measuring up to 0.3-0.4 cm in diameter and usually with a chlorotic zone around the spot.

The fungus was identified microscopically and confirmed by its morphological characterization and gave the name of *Alternaria solani*. The sequences of the isolate studied were compared with those of NCBI databases using BLAST network. The partial sequences of 18S rRNA obtained from *Alternaria* isolate was aligned with the available 18S rRNA sequences in Gene Bank data base, and the fungus was found to have 99-100% similarity with *A. solani* strains (Table 1).

Partial sequencing of 18S and 28S rRNA and complete sequence of internal transcribed sequence 1 (ITS1), internal transcribed sequence 2 (ITS2) and 5.8S rRNA gene has

been submitted in Sigma Aldrich Company (Fig. 1). Multiple sequences alignments and phylogenetic tree illustrated graphically in Fig. 2.

3.2 Effect of Kocide Fungicide on the Growth of *A. solani*

The control of plant diseases is one of the most important topics. Scientists all over the world are trying to solve the complicating problems of plant diseases. The control of plant diseases has been possible by the use of chemicals or fungicides.

Data in Fig. 3 showed that, the growth of *A. solani* was completely inhibited at 600 ppm of Kocide, while at 50 up to 250 ppm has no effect against the same fungus and starting from 400 ppm; the inhibition was increased gradually up to 600 ppm which gives complete inhibition (100%).

Table 1. Sequence producing significant alignments for *A. solani* isolate

Accession No.	Description	Identity %
KC78609.1	<i>Alternaria solani</i> strain UP T 10	100%
KJ634071.1	<i>Alternaria solani</i> strain SL 1	100%
KF998549.1	<i>Alternaria solani</i> strain 043021	100%
AY154716.1	<i>Alternaria solani</i> strain IA300	100%
AY154715.1	<i>Alternaria solani</i> strain IA295	100%
KF308830.1	<i>Alternaria solani</i> isolate IM HW A6	100%
KF998550.1	<i>Alternaria solani</i> strain A7AXTL14e 2	100%
AF314576.1	<i>Alternaria solani</i> strain AS3	99%
JQ625583.1	<i>Alternaria solani</i> f sp lycopersici isolate UP 7	99%
JQ625580.1	<i>Alternaria solani</i> f sp lycopersici isolate HR 12	99%
JF331442.1	<i>Alternaria porri</i> isolate CNU093037	99%
JQ625586.1	<i>Alternaria solani</i> f sp lycopersici isolate RJT 1	99%
JX418359.1	<i>Alternaria potato</i> strain A38	99%

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GGGCTTGGTGGATGCTAGACCTTGGGGCTGGAAAGAGAGCGCGACTTGTGCTG
GAAAAAAAGGCTTGGTGGATGCTAGACCTTGGGGCTGGAAGAGAGCGCGACTTGTGCTG
CGCTCCGAAACCAGTAGGCCGGCTGCCAATGACTTTAAGGCGAGTCTCCCGCAAAGGGAG
CGCTCCGAAAACAGTAGGCCGGCTGCCAATGACTTTAAGGCGAGTCTCCCGCAAAGGGAG
ACAAAAAGACGCCCAACACCAAGCAAAGCTTGAGGGTACAATGACGCTCGAACAGGCAT
ACAAAAAGACGCCCAACACCAAGCAAAGCTTGAGGGTACAATGACGCTCGAACAGGCAT
TGCAATTCACACTACTTATCGCATTTTCGCTGCGTTCATCGATGCCAGAACCAAGAGA
TGCAATTCACACTACTTATCGCATTTTCGCTGCGTTCATCGATGCCAGAACCAAGAGA
TCCGTTGTTGAAGTTGTAAATTAATACATTTGTTA CTGACGCTGATTGCCATTGCAAAAA
TCCGTTGTTGAAGTTGTAAATTAATACATTTGTTA CTGACGCTGATTGCCATTGCAAAAA
GGTTTATGGGTTGGTCTTGTGGTGGGCGAGCCCAAGGAAACAAGAAGTACGCAAAAA
GGTTTATGGGTTGGTCTTGTGGTGGGCGAGCCCAAGGAAACAAGAAGTACGCAAAAA
GACACGGGTGAATAATTCAGCAAGGCTGGCCACCCGGGAGGTGCCAGCCCGCCTTCAT
GACACGGGTGAATAATTCAGCAAGGCTGGCCACCCGGGAGGTGCCAGCCCGCCTTCAT
ATTTGTGTAATGATCCCTCCGAGGTTCACTACGGA
ATTTGTGTAATGATCCCTCCGAGGTTCACTACGGA
    
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Fig. 1. *A. solani* isolate sequence based on partial 18S rRNA sequences

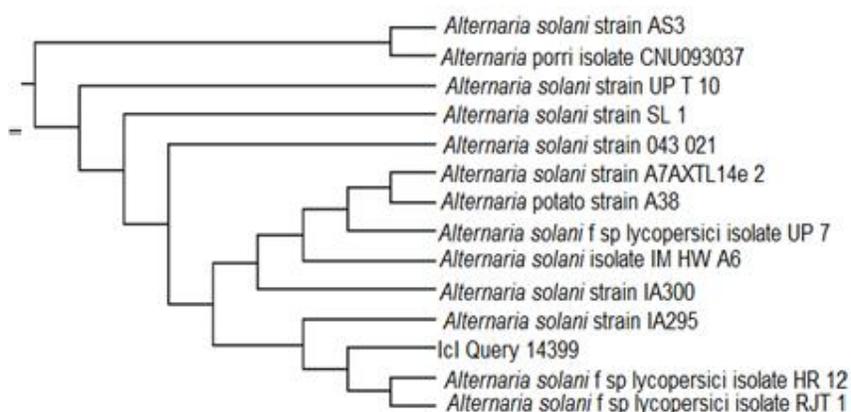


Fig. 2. Phylogenetic tree showing the position of isolate *A. solani* derived from 18S rRNA gene sequences

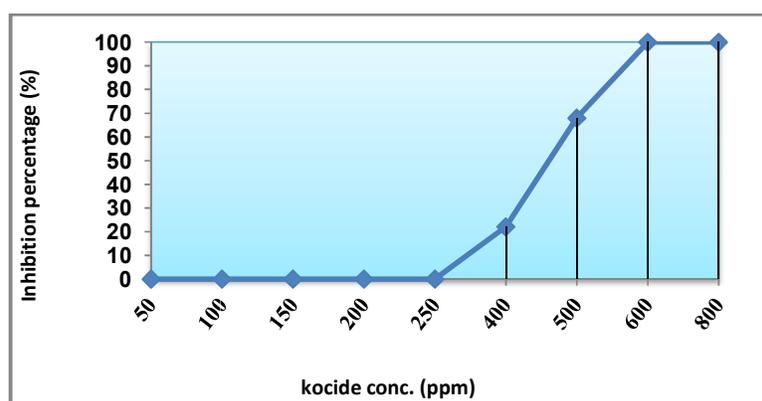


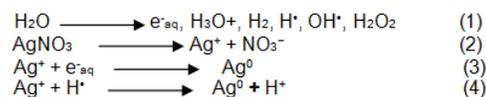
Fig. 3. Effect of Kocide on the growth of *A. solani* *in vitro* after 7 days

Fungicides can be divided into two groups based on mode of action in fungal cells: a. Site-specific inhibitors: Site-specific inhibitors target individual sites within the fungal cell. b. Multi-site inhibitors that target many different sites in each fungal cell. Where, many fungicides in the market having the same mode of action where they are sterol and demethylation inhibitors or interfering with respiration and interruption of energy compound as ATP production in sensitive fungi [20]. Also, as reported by Meena et al. [21] fungicides mancozeb and carbendazim caused 100% reduction in mycelial growth of *Alternaria brassicae* over control *in vitro*.

3.3 Synthesis of Ag Nanoparticles using Gamma Irradiation

A method using gamma radiation provides more convenient and a cleaner approach. AgNO_3 separated to Ag^+ and NO_3^- ions in the aqueous

solution as shown in Equation 2. Gamma radiolysis of aqueous solution yields several products including hydrated electrons and hydrogen atoms (H^\bullet) having the powerful reducing ability that reduce Ag^+ into Ag^0 [22].



The growth of silver nanoparticles by reduction of Ag^+ to Ag^0 is stepwise. These neutral Ag^0 atoms at first dimerize when they encounter or associate with the excess Ag^+ ions trapped in the individual loops of cell free supernatant. The charged dimer clusters Ag^{2+} may further react with excess silver cations by a -cascade of coalescence processes to form trimer, tetramer and higher order silver ion clusters (Ag^{n+}) and also the doubly charged 3Ag^{2+} , 4Ag^{2+} , etc. The aggregation of these clusters into higher metallic

clusters and nanoparticles occurs as the nucleation in the solution increases. The competition between the reduction of free silver ions and absorbed ones is controlled by the rate of reducing radical formation [23]. Silver atoms formed by the irradiation tended to coalesce into oligomers (Equation 5), which progressively grew into large clusters (Equation 6). The aqueous electrons reacted with the Ag^+ clusters to form the relatively stabilized Ag clusters (Equation 7) as illustrated in [24].



Fig. 4 showed that, after gamma irradiation of mixture of *T. viride* supernatant and $AgNO_3$ from 0.5 up to 30 kGy compared with unirradiated control. For all doses, the characteristic surface Plasmon resonance peak for silver nanoparticles appear, that is indication

of AgNPs formation while, the maximum AgNPs production was at 15 kGy. On the other hand, over than 15 kGy the production of AgNPs was decreased.

3.4 Characterization of Silver Nanoparticles Synthesized by Gamma Irradiated *T. viride* Supernatant

Average particle size was determined by DLS. Fig. (5a) represented graph of AgNPs solution exhibit a very narrow size distribution with small particles size. The volume distribution of the hydrodynamic size of the nanoparticles show peak (approximately 100% of the particle volume) had its average size at 14.5 nm. TEM image of AgNPs solution with the average size of nanoparticles to be 12.7 nm Fig. (5b). The nanoparticles were monodispersed without aggregation indicating stabilization of the nanoparticles by a protein capping agent.

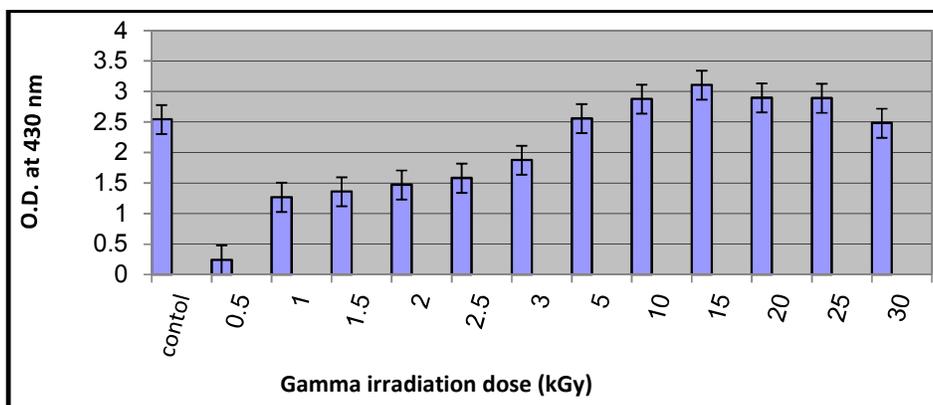


Fig. 4. Optical density of AgNPs after gamma irradiation at different doses

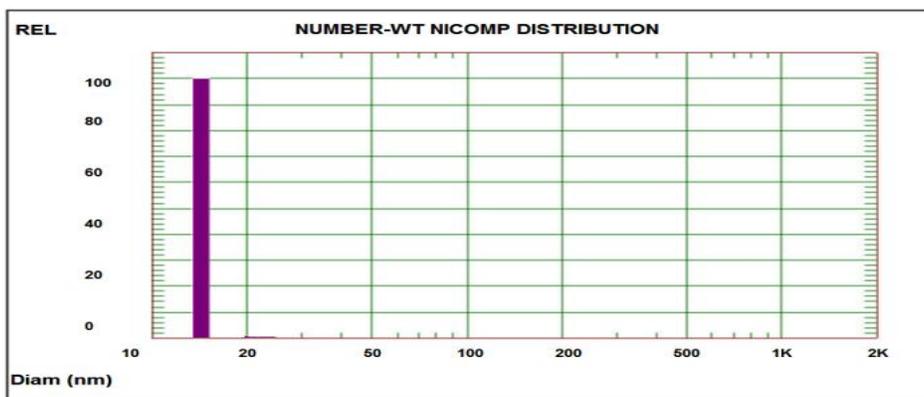


Fig. 5(a). DLS (mean diameter = 14.5 nm, Num = 100 %)

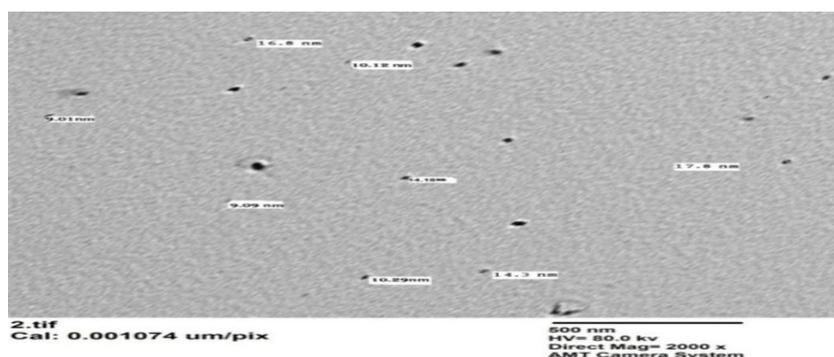


Fig. 5(b). TEM graphs (average diameter of 12.7 nm) of AgNPs synthesized by *T. viride* cell free supernatant

3.5 Effect of Silver Nanoparticles on the Growth of *A. solani*

Applications of nanoparticles based fungicides have been increasing gradually in recent years because of many advantages over conventional chemical fungicides. This study presents fungicidal efficiency of silver nanoparticles against the phytopathogen, *A. solani*.

Data in Fig. 6 showed that, the growth of *A. solani* was completely inhibited at 25 ppm AgNPs, while at 5 ppm the lowest effect against the same fungus was obtained. AgNPs were highly effective against the pathogenic fungus *A. solani* *in vitro*.

In previous studies it was observed that, silver nanoparticles disrupt transport systems including ion efflux. The dysfunction of ion efflux can cause rapid accumulation of silver ions, interrupting cellular processes at their lower concentrations such as metabolism and respiration by reacting with molecules. Also, silver ions are known to

produce reactive oxygen species via their reaction with oxygen, which is detrimental to cells, causing damage to proteins, lipids, and nucleic acids [25,26].

Different parameters, such as particle size and particle concentration, which affect the efficiency of the fungicidal effect, showed that small sized particles (12.7 nm) are very effective in preventing the fungal growth. It is well known that silver nanoparticles are highly toxic to microorganisms. Silver nanoparticles have been known to have inhibitory and antimicrobial effects and thus we extend its application as an antimicrobial agent. Several studies propose that silver nanoparticles may attach to the surface of the cell membrane disturbing permeability and respiratory function of the cell. It is also possible that silver nanoparticles not only interact with the surface of membrane, but can also penetrate inside the organism [27]. In addition, Naghsh et al. [28] found that growth and colony number of *Aspergillus niger* decreased significantly in 12.5 ppm concentration of nano-silver.

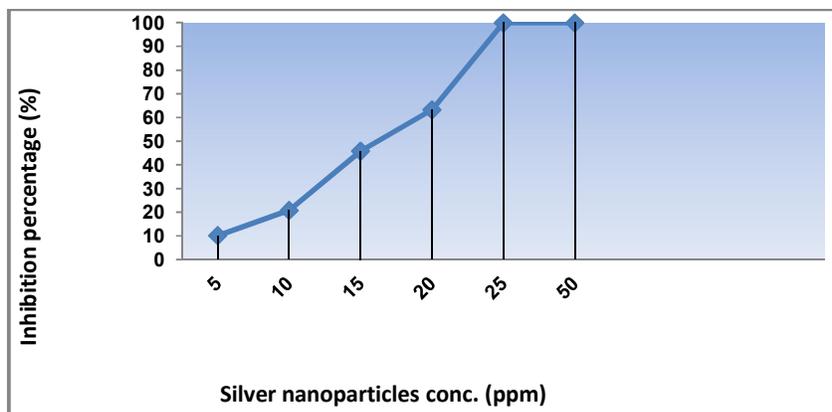


Fig. 6. Effect of AgNPs on the growth of *A. solani* *in vitro* after 7 days

Silver stops the metabolism of pathogenic microorganisms. The electrical load of an emitted silver anion kill the pathogenic microbes by controlling a procreation function [29,30]. It was found that there are no bacteria which can survive after touching silver for 6 minutes [31]. These results suggested that, nanosilver is a potential compound in the treatment of fungal infectious diseases. Many studies have shown the antimicrobial effect of nanosilver [32]. The same results was recorded with Gajbhiye et al. [33] who observed that, the transmission electron microscopy analysis revealed the interaction between nano-Ag and the membrane structure of *Candida albicans* cells during nano-Ag exposure results in changes in the membranes of *C. albicans*, which can be observed as the "pits" on the membrane surfaces. The formation of pores subsequently leads to cell death. In this mechanism, the nano-silver particles diffuse gradually silver ions. These ions substitute the -SH bonds to -SAg ones in microorganism's membrane in a substitution reaction, resulting in death of microorganism [34]. Rathod et al. [35] studied synthesizing highly stabilized monodispersed silver nanoparticles by *Rhizopus stolonifer* and the antifungal efficacy of silver nanoparticles (AgNPs) against *Candida* sp. Also, Lamsal et al. [36] applied silver nanoparticles at various concentrations. Silver nanoparticles caused a detrimental effect on mycelial growth of *Colletotrichum* species *in vitro* and in the field. Molecular mechanism provided by silver nanoparticles can be attributed to production of free radicals. These free radicals were induced oxidative stress and program cell death in microorganisms. These free radicals result in apoptosis, Programmed death of live cells, by attacking the intracellular organelles of fungi [37].

3.6 Effect of Selenium on Pathogenic Fungus, *A. solani* *in vitro*

As shown in Fig. 7 selenium nanoparticles exert gradual inhibiting effect on the growth of the pathogenic fungus, *A. solani*. The lowest concentration 50 ppm gave 8.33% of inhibition while, at 800 ppm of selenium nanoparticles completely inhibit the fungus growth (100% inhibition). *A. solani* tolerate high concentrations of SeNPs.

Many literatures attributed the antibacterial effects of different selenium compounds to the formation of free radicals [38]. Moreover, selenium has also been found to trigger the generation of ROS, with both elements capable of reacting with intracellular thiols and forming intermediates that cause oxidative stress as a consequence of the formation of superoxide radicals [39,40]. Nanoparticles can contribute to functional damages of cell membrane or wall by disrupting the integrity of these important envelopes [41].

At similar research, Khalifa and Sameer [42] indicated that, selenium started to inhibit the fungal growth of *Penicillium digitatum*, causing green mold disease of orange fruit growth at 100 ppm with 5.56% inhibition. However, at 500 ppm, selenium gave fungal growth inhibition rate of 85.22%. Also, Sameer [43] found that, *Alternaria solani* tolerated high concentrations of selenium according to its IC_{50} value (263.03 ppm). In this respect Zohri et al. [44] reported that, the fungal biomass of *Aspergillus parasiticus* was slightly decreased at higher concentrations of selenium.

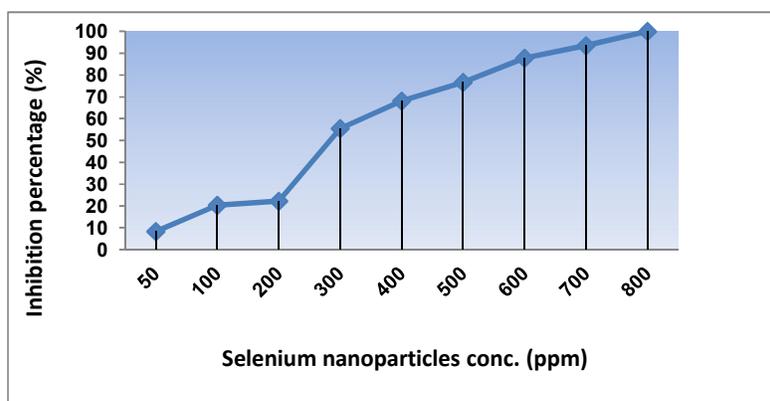


Fig. 7. Effect of SeNPs on the growth of *A. solani* *in vitro* after 7 days

4. CONCLUSION

In the present work it was demonstrated that, silver nanoparticles has significant antifungal activity against potato pathogenic fungus, *A. solani*. So, AgNPs can be effectively used as antimicrobial agents to control early blight disease of potato. SeNPs have a moderate effect on *A. solani*. So, AgNPs may be applied widely and safely instead of using the commercially available synthetic fungicides, which show higher toxicity to humans.

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COMPETING INTERESTS

Authors have declared that no competing interests exist.

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