Asian Journal of Environment & Ecology



14(3): 36-43, 2021; Article no.AJEE.65955 ISSN: 2456-690X

Evaluation of the Efficiency of Ceramic Water Filter Improved by Dried Duckweed Plant (*Lemna minor*) in Wastewater Treatment

S. A. Osemeahon¹, J. O. Okechukwu¹ and B. J. Dimas^{2*}

¹Department of Chemistry, Modibbo Adama University of Technology, Adamawa State, Nigeria. ²Department of Science Education, Taraba State University, Jalingo, Nigeria.

Authors' contributions

This work was carried out in collaboration among all authors. Author JOO designed the study, performed the statistical analysis, wrote the protocol and wrote the first draft of the manuscript. Authors SAO and BJD managed the analyses of the study and supervised the work. Author BJD managed the literature searches. All authors read and approved the final manuscript.

Article Information

DOI: 10.9734/AJEE/2021/v14i330209 <u>Editor(s):</u> (1) Prof. Daniele De Wrachien, State University of Milan, Italy. <u>Reviewers:</u> (1) Kobilov Aziz Mukhtorovich, Bukhara State University, Uzbekistan. (2) Xiaoting Li, Sichuan Normal University, China. (3) Rodrigo Vieira Blasques, Federal University of São Carlos, Brazil. Complete Peer review History: <u>http://www.sdiarticle4.com/review-history/65955</u>

> Received 10 January 2021 Accepted 15 March 2021 Published 15 April 2021

Original Research Article

ABSTRACT

The lack of clean water sources due to pollution and industrialisation is a major problem in many countries including Nigeria. To overcome this challenge, various methods have been adopted including phytoremediation treatment. This study evaluates dried duckweed an aquatic plant and its removal efficiency in comparison with other locally available treatment materials. This was achieved by formulating ceramic water filters (C.W.F) categorized into four different types- clay and kaolin(P1), clay, kaolin and sawdust(P2), clay, kaolin and charcoal(P3) and clay, kaolin and duckweed(P4). These filters were subjected to contaminated water and the following physicochemical parameters Colour, pH, Conductivity(Ec), Fluoride(F⁻), Magnesium(Mg²⁺), Nitrites(NO₂⁻), Sulphates (SO₄²⁻), Ammonia (NH₃) and Total Suspended Solids (TSS) and Total Nitrogen were determined before and after filtration. In all the ceramic water filters, the filter improved by duckweed showed the best removal efficiency of Colour – 100%, Conductivity(Ec) - 72.60%, Fluoride(F) - 99.82%, Magnesium(Mg²⁺) - 51.68% Nitrites(NO₂⁻)-92.34, Sulphates (SO₄²⁻)-

^{*}Corresponding author: E-mail: blesseddimas@yahoo.com;

46.09%, Ammonia (NH_3)-98.75%, and Total Suspended Solids (TSS)- 85.43% and Total Nitrogen (TN) -83.79% indicating that duckweed is capable of adsorbing inorganic and organic pollutants from water.

Keywords: Ceramic water filters; clay; duckweed; phytoremediation; water sources; water treatment.

1. INTRODUCTION

Availability of fresh water controls the major part of the world economy. The adequate supplies of water are necessary for agriculture, human consumption, industry as well as recreation. With the advancement of technology and industrial growth, fresh water resources all over the world are being threatened. One sixth of the world's population suffers from fresh water unavailability situation [1].

Contaminated water causes problems to health and leads to waterborne diseases which can be prevented by taking measures even at the household level. In both developing and industrialized nations, a growing number of contaminants are entering water supplies from human activity: from traditional compounds such as heavy metals and distillates to emerging micro-pollutants such as endocrine disrupters and nitrosamines. Increasingly, public health and environmental concerns drive efforts to decontaminate waters previously considered clean. More effective, lower-cost, robust methods to disinfect and decontaminate waters from source to point-of-use are needed, without further stressing the environment or endangering human health by the treatment itself [2].

Effective water treatments through conventional methods which rely on heavy aeration are expensive to install and operate. Hence, there is a need to explore some non-conventional methods which are not only economically viable and easy to operate but are eco-friendly [3].

For this purpose, duckweed *lemnar minor* a floating macrophyte as shown in Fig. 1, serves as phyto-remediation technology which is capable of extracting nutrients or pollutants coupled with a fast growing rate proves to be an excellent candidate for water treatment. Duckweed grows faster when the nutrients are in abundant medium. They utilize waste water nutrients and are cost effective, yielding protein-rich plant biomass as by-products [4].

Also, duckweeds possess a unique ability to accumulate starch and this can be converted into

ethanol as a source of bio-fuel. Moreover, they can grow on less fertile land and hence they do not compete for fertile land that can be used for growing other agricultural commodities [5].



Fig.1. Duckweed plant (Lemna minor)

Duckweed wastewater treatment systems remove, by bioaccumulation, as much as 99 percent of the nutrients and dissolved solids contained in wastewater [6]. These substances are then removed permanently from the effluent stream following the harvesting of a proportion of the crop. The plants also reduce suspended solids and biological oxygen demand (BOD) by reduction of sunlight in lagoons. Duckweed systems distinguish themselves from other effluent wastewater treatment mechanisms in that they also produce a valuable, protein-rich biomass as a by-product. Depending on the wastewater, the harvested crop may serve as an animal feed, a feed supplement supplying protein/energy and minerals, or a fertilizer. The question of toxic elements must be considered if certain types of waste material serve as the nutrient source for duckweed culture; for example, duckweed will absorb heavy metals and insecticides from the wastewater. It may, therefore, have to be decontaminated prior to feeding to animals if heavy metals are present in the water [3].

Most researchers, however, suggest that efficiency gains using duckweed are greater in secondary and tertiary treatments of effluents where organic sludge has already been removed or converted into simple organic and inorganic molecules that can be used directly by duckweed [7,8,9,10].

The performance of ceramic water filters has been significantly improved by the use of burnout materials which increase the flow rate by creating a network of pores. Burn out materials such as sawdust helps generate pores, increasing the flow rate and reducing the efficiency of the ceramic water. Charcoal which has been used as an additional material to adsorb odour and colour from the filter hence, further increasing the flow rate with no significant removal of inorganic and organic pollutants which are very detrimental to health [11].

This study evaluates the potentials of dried duckweed plant, formulated and incorporated into a ceramic water filter, and its effectiveness in treating water when compared with other locally available materials. To achieve this, four (4) different filters were produced containing clay and kaolin (P1) clay, kaolin and sawdust (P2) clay, kaolin and charcoal (P3) and clay, kaolin and duckweed (P4). These filters were treated with contaminated water and various physico-chemical properties of the water were measured before and after treatment.

2. MATERIALS AND METHODS

The materials used for the research are: clay, kaolin (AL₂O_{3.}2SiO_{2.}2H₂O), sawdust, charcoal, duckweed, and water. The clay was obtained from Girei, Adamawa State. Kaolin was obtained from Alkaleri LGA of Bauchi State. Sawdust and charcoal were obtained in Jambutu, Adamawa State. Duckweed plants were obtained from Lake Gerio, Yola, Adamawa State, Nigeria.

2.1 Formulation of the Ceramic Water Filters

The different mixtures of the filters were moulded by ceramists from Industrial Design and Ceramics Department in MAUTECH, Yola. The mixtures were moulded into a filter of clay and kaolin (P1), clay, kaolin and sawdust (P2), clay, kaolin and charcoal (P3) clay, kaolin and Duckweed (P4) as seen in Fig. 2 and 3. The different moulded samples were dried for four (4) weeks to remove all moisture content and fired to prevent cracking as seen in Fig.4. This further reduced the size of the ceramic filters [13].



Fig.2. Mould for casting formulations



Fig.3. Formulated filter "green" wares



Fig.4. Fired ceramic filters

Treatment	Availability & practicality	Technical difficulty	Cost	Microbial efficacy
Boiling at 100°C	Varies	Low-moderate	Varies	High
Chemical treatment	High to moderate	Low-moderate	Moderate	High
Solar disinfection	High	Low-moderate	Low	Moderate
UV lamp treatment	Varies	Low-moderate	Moderate– High	High
Coagulation sedimentation/filtration	Varies	Low-moderate	Varies	Varies

Table 1. General properties of house water treatment system

Source [12]

2.2 Filtration Test

The efficiency of the filtration by each filter indicates that the filters must be able to reduce as much contaminants as possible to a level within the standard drinking water range. Hence, the water before the filtration was from a wellknown contaminated source.

The source of the contaminated water as seen in Fig. 11 was collected from Bwarranji community which gets its drinking and domestic water from the off stream of River Benue which covers about 1,083km in length and rises in Northern Cameroun as B'enoue at about 1,340m and it is located in Adamawa State a North-eastern part of Nigeria with a population of 3,737,223 people and land mass of 36,917km² and it supplies water to the treatment plant in Jimeta-Yola the Adamawa capital which is located between Longitudes 12°26' E and Latitude 9° 16 N [14] and serves as a source of livelihood for the local community.

The contaminated water was obtained using the 20litres rubber bucket and it was visibly very dirty as seen in Fig 5. It was directly filtered the ceramic water filters within 24 hours. This was done by tightly placing each of the ceramic water filters on a 13litres transparent rubber as seen in Fig 5. The filter pots filtered the contaminated water for 24 hours so as to get appreciable quantity of water for the analysis as seen in Fig 8. The rubbers were labelled "P1" "P2" "P3" "P4" respectively for different filter pots with such label.



Fig.5. Sample of bad water



Fig.6. Filtration of contaminated water



Fig.7. Filtration Process



Fig.8. Filtration set-up



Fig.9. Filtrates from different CWF

2.3 Water Analysis

The following parameters were analysed: Odour, Colour, pH, Conductivity (Ec), Fluoride (F^{-}), Magnesium (Mg²⁺), Nitrites (NO₂⁻), Sulphates (SO₄²⁻), Free Ammonia (NH₃), and Total suspended solids (TSS).

The samples of the filtered water (after) and the unfiltered water (before) were collected in a clean wash bottle. The sample was first used to rinse the wash bottles before filling [15].

The sample wash bottles were labelled "before" "P1" "P2" "P3" "P4" for the contaminated water (before) and filtered water (after) respectively. The wash bottles containing each sample of water was taken within 24 hours to the Adamawa State Water Board Chemical Laboratory for analysis.

3. RESULTS AND DISCUSSION

The texture of the ceramic filters as seen in table2 Shows that P1 had a smooth texture because there was no addition of sawdust. While P2 texture was significantly rough due to the addition of sawdust.

However, this roughness reduced from P2 to P3 to P4, and this can be attributed to the decrease in the percentage of burn-out materials in the filter formulation.

In table3 it was observed that the removal efficiency of the filters with respect to anions (F^{-}), cations (Mg²⁺), nitrites (NO₂⁻), sulphates (SO₄²⁻), and free ammonia (NH₃), was due to the ion exchange on the ceramic surface and formation

of precipitate as oxides and hydroxides, and only P4 filter which a gave a reduction value of 51% agreed to the study by [16], that most developed ceramic filters could remove cations with more than 50% from water. However, the removal efficiency of magnesium and sulphate ions was apparently low thus increasing the total hardness (MgSO₄) of the filtered water. This might be due to the high weathering activities of the rocks around the community, which indicates the presence of the mineral, gypsum [17]. According to Nigerian Industrial Standards [18] and World Health Organisation [19] report, there is no health implication for maximum presence of hardness. However, P4, showed the highest removal efficiency of these anions and cationic radicals due to its high adsorption capacity [16].



Fig.10. Map of Nigeria showing River Benue Source: en.wikipedia.org/wiki/Benue_River



Fig.11. Water sample collection point at Bwarranji community an off-stream of River Benue Source: www.gettyimages.com

Filter	Texture/Appearance		
P1	Smooth		
P2	Highly rough		
P3	Moderately rough		
P4	Slightly Rough		

Table 2. Texture/Appearance of Each filter

S/N	Parameters	Before Treatment	After Treatment			
			P1% P2%	, 0	P3%	P4%
1	Colour(TCU)	2.89	100	99.27	99.41	100
2	pH	9.64	7.86	8.00	8.15	7.49
3	Conductivity(mg/l)	520.0	70.29	52.69	62.60	72.60
4	Fluoride(mg/l)	2.23	99.69	99.51	99.60	99.82
5	Magnesium(mg/l)	8.24	38.72	42.21	37.38	51.68
6	Nitrites(mg/l)	0.257	100	100	100	100
7	Sulphates(mg/l)	29.77	40.64	39.37	39.67	46.09
8	Ammonia(mg/l)	16.42	94.21	93.89	93.61	98.75
9	TSS(mg/l)	240.00	73.78	65.40	73.73	85.43
10	TN(mg/l)	37.32	78.27	72.96	75.54	83.79



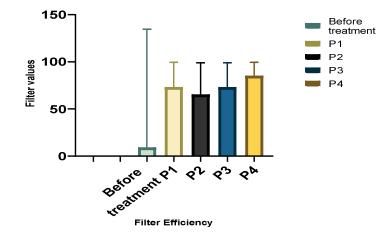


Fig.12. Bar graph showing the median values and interquartile range of the different CWF

The ceramic filters were evaluated for conductivity (Ec) and pH of the source water. Electrical conductivity is related to the amount of ions present in a given solution. The conductivity was significantly reduced by P4 (72.60%) filter because of its high adsorption capacity of these ions, however, this shows that the turbidity of the water is high.

The pH value of the source water (9.64) was relatively high in alkalinity and was above the maximum permissible limit (6.5-8.5) prescribed by Nigerian Industrial Standard [18] but was reduced after filtration by all the filters within a range of (7.86-8.15) as seen in Table 3. The decrease in pH of the filtrate indicates a significant reduction of the Total Suspended Solids by adsorption [20].

The average reduction of ammonia by P4 was 98% and this was slightly above the value reported by Selavarani et al. [21] which was 96% but agree to the value reported by Kutty et al. [22] which was 98% indicating that duckweed gives a significant reduction of ammonia. Removal efficiency of 99% of nitrite was also

reported by Selavarani et al. [21] and this agrees with our result for all the filters (>99%) for nitrite removal. A significant reduction of fluoride by all the filters was within 99% range and agrees with the report of Kiaglo et al. [23] which was 99.33%, indicating that addition of burn out material increases the removal efficiency of fluoride in wastewater.

The average reduction of TN for all the filters was >70% except for P4 which was 84%, this value was higher than the value reported by Mohammed and Rasha [24] which was 63%. This shows that the filters significantly reduced the decomposed inorganic and organic aerobic nitrogenous matter from the wastewater.

Total Suspended Solid (TSS) is the measure of particulate matter suspended in water and it is used to describe the extent of pollution in wastewater [25]. According to our study, P4 gave the highest removal efficiency of 85.43% and this agrees with the study made by Gupta [3]. The lowest removal efficiency was recorded by P2, this was due to the addition of sawdust in its formulation, hence increasing its pore size and allowing suspended solids to pass through. The low pore size recorded for P1, P3 and P4 enables the filters to remove particles and pathogens by size exclusion principle [26].

Fig. 8. shows the median and interquartile range of values for filters both in the treatment of the different water parameters and between the filters. From the 2 way ANOVA analysis there was a significant difference (P=0.0021) between P1, P2, P3 and P4. A significant difference (P< 0.0001) was recorded between the water parameters (treatment) and the filters (exposure).

4. CONCLUSION

The result presented here are the most recent systematic comparison of dried duckweed composite filters and their relative removal efficiency of pollutants and contaminants from wastewater. Our key finding from the analyses is that dried duckweed is as effective as its fresh state in water and can improve CWFs in wastewater treatment. This confirms the studies made by most researchers that duckweed plant is a good phytoremediation tool for wastewater treatment. This suggest that the quality of drinking water especially in developing countries like Nigeria, can be enormously improved by the use of duckweed, either incorporated as ceramic water filter, duck weed pond or duckweed tank.

Duckweed which have been under-utilized have proven to be an efficient nutrient sink and an effective tool in contaminated water treatment. The use of this rich aquatic plant will go a long way to reduce the burden of water- borne diseases and also ensure safe, clean and sustainable water supply.

ACKNOWLEDGEMENT

The authors acknowledge the Department of Chemistry and Industrial Design and Ceramics, Modibbo Adama University of Technology, Yola for their assistance and support.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

REFERENCES

- 1. Elimelech.The global challenge of adequate and safe water. Journal of Water Supply Resource Technology AQUA. 2006; 55:3-10.
- Shannon MA, Bohn PW, Elimelech M, Georgiadis JG, Marinas BJ, Mayes AM. Science and technology for water purification in the coming decades. Nature. 2008; 452:301-310.
- 3. Gupta R, Prakash P. Duckweed: An Effective Tool for Phyto-remediation, Amity University, Noida, India; 2014. DOI: http://dxidoi.org/10.1080/027722482012.87

http://dx:doi.org/10.1080/027722482013.87 9309

- 4. Ferodoushi L, Haque F, Khan S, Haque MM. The Effects of two Aquatic Floating Macrophytes (*Lemna* and *Azolla*) as biofilters of nitrogen and phosphate in fish ponds. Turkish Journal of Fisheries and Aquatic Sciences. 2008;8(2):253-258.
- 5. Cheng JJ, Stomp AM. Growing duckweed to recover nutrients from waste water and for production of fuel ethanol and animal feed. Clean-Soil, Air, Water. 2009;37:17-26
- Patel DK, Kanungo VK. Phytoremediation potential of duckweed (*Lemna minor* I. A tiny aqautic plant) in the removal of pollutants from domestic waste water with special reference to nutrients. The Bioscan International Journal of Life Science 2010;5(3):355-358
- Alaerts GJ, Rehman M, Kelderman P. Performance Analysis of a full scale duckweed covered lagoon. Water Research. 2006;30:843-852

- Caicedo JR, Van Der Steen, Arce O, Gijzen H.J. Effect of Total Ammonia Nitrogen Concentration and pH on Growth Rates of Duckweed (*Spirodela polyrrhiza*). Water Research 2000;34:3829-3835
- 9. Smith MD, Moelywati. Duckweed Based Wastewater Treatment (DWWT): Design guidelines for hot climates water science and technology. 2001;43:291-299.
- Dalu JM, Ndamba J. Duckweed based wastewater stabilisation ponds for wastewater treatment (a low cost technology for small urban areas in Zimbabwe). Physics and Chemistry of Earth, Parts A/B/C. 2003; 28:1147-1160.
- Ajayi BA, Lamidi YD. Formulation of ceramic water filter composition for the treatment of heavy metal and correction of physiochemical parameters in household water; Art and Design Review. 2015;(3):94-100.
- Annan E, Benjamin A, Yaw DB, David SK, Abu Y, Boateng O. et al. (2018) Application of Clay Ceramics and Nanotechnology in Water Treatment: A Review, Cogent Engineering. 2018;5:1476017.
- Nnaji CC, Afangideh BC, Ezeh. Performance of Clay-Sawdust Composite Filter for Point of Use Water Treatment Nigerian Journal of Technology (NIJOTECH). 2016;35(4):949-956
- Hong AH, Law PL, Selaman OS. Physicochemical quality assessment of pollutants in river benue water in Jimeta/Yola Metropolitan, Adamawa State, Northeastern Nigeria. Am. J. Environ. Prot 2014;3(2):90-95.
- 15. APHA. Standards Methods for the Examination of Water and Waste Water, 22nd Edition; 2012.
- 16. Enyew AZ, Bekalo TB. Clay ceramic for water treatment. Materials Science and Applied Chemistry Journal 2017; 34:69-74.
- 17. Okechukwu JO, Osemeahon SA, Dimas BJ. Development and Evaluation of

Ceramic Water Filter Prepared from Dried Duckweed Plant. African Journal of Environmental and Natural Science Research 2021;4(1):53-61.

- Nigerian Industrial Standard. Nigerian Standard for Drinking Water Quality; 2015. ICS 13.060.20 NIS-554-2015
- 19. World Health Organisation. Guideline for Drinking Water Quality 3rd Edition; 2004.
- Werkeneh AA, Medhanit BZ, Abay KA, Dante YJ. Physico-chemical analysis of drinking water at Jigjiga City, Ethiopia. American Journal of Environmental Protection. 2015;4:29-35.
- Selvarani JA, Padmavatly P, Srinivasan A, Jawahar P. Performance of duckweed (*Lemnar minor*) on different types of wastewater treatment. International Journal of Fisheries and Aquatic Studies. 2015;2(4):208-212
- 22. Kutty SRM, Ngatenah MH, Malak Ahmed A. Nutrient Removal from Municipal Wastewater Treatment Plant Effluent using *Eichhornia crassipies*, World Academy of Science, Engineering and Technology 2009;6:826-831.
- 23. Kiaglo B, Revocatus M, Askwar H, Karoli N. Performance of water filters towards the removal of selected polluntants in Arusha, Tanzania.Tanzanian Journal of Science 2016;42:134-147.
- 24. Mohammed A.A. and Rasha A.J. Treatment of domestic wastewater using duckweed plant. Journal of King Saud University- Engineering Sciences. 2010; 22(1):11-18.
- Tolulope EA, Temilola O, John OO, Joshua NE. Physico-chemical analysis of wastewater discharge from selected paint industries in Lagos Int. J. Environ. Res. Public Health 2019;16(7):1235.
- Bielefeldt AR, Kowalski K., Schilling C, Schreier A, Scott Summers R. (2010) Removal of virus to protozoan sized particles in point-of use water filters, Water Research, 2010;44(5):1482-1488.

© 2021 Osemeahon et al.; This is an Open Access article distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

> Peer-review history: The peer review history for this paper can be accessed here: http://www.sdiarticle4.com/review-history/65955